**NOMENCLATURE REVIEW**

**Molecular Compounds, Ionic Compounds, & Acids**

NAME THE FOLLOWING COMPOUNDS:

1. BaSO3 14. AgNO3
2. (NH4)3PO4 15. As2O5
3. PBr5 16. Fe2O3
4. MgSO4 17. HClO
5. CaO 18. N2O3
6. H3PO4 19. HF
7. Na2Cr2O7 20. H2C2O4
8. MgO 21. NaHCO3
9. SO3 22. SiBr4
10. Cu(NO3)2 23. CuCl2
11. HI 24. HNO2
12. N2O 25. SnO2
13. MnO 26. BaCrO4

WRITE FORMULAS FOR THE FOLLOWING COMPOUNDS:

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| 1. hydrobromic acid 2. chromium(III) carbonate 3. magnesium sulfide 4. iodine trichloride 5. lithium hydride 6. ammonium hydroxide 7. calcium chloride 8. hydroselenic acid 9. iron(II) nitride 10. aluminum hydroxide 11. tin(II) fluoride 12. sulfur tetrachloride 13. mercury(II) iodide | 1. diphosphorus pentoxide 2. sulfurous acid 3. lead(II) nitrate 4. dihydrogen monoxide 5. sodium oxalate 6. perchloric acid 7. chlorous acid 8. silicon dioxide 9. carbonic acid 10. sodium chlorate 11. xenon hexafluoride 12. nickel nitrate 13. potassium perchlorate |